REACTION OF TETRACHLORO-O-BENZOQUINONE WITH 2-VINYLFURANS

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(Received in UK 16 June 1969; accepted for publication 25 August 1969)

During the course of our studies on quinonoid compands, it has been found that tetrachloro-o-benzoquinome (I) has pronounced cytostatic activity. This stimulated us to investigate its reactivity toward the known cytostatic agent, 1,1-dicyano-2-(2-furyl) ethylene (IIa)¹. It is found that (I) adds preferentially to the double bond of the furyl residue to give the 1,4-dio-xen (IIIa) leaving the 2-vinyl side chain intact. It adds similarly to the analogous 2-vinylfurans (IIb-e) in boiling benzene to give the corresponding 1,4-dioxen derivatives, (IIIb-e). Thus, in this respect the isolated furan

$$\begin{array}{c} \text{C1} & \text{C1} & \text{C1} & \text{C1} \\ \text{C1} & \text{C1} & \text{C1} & \text{C1} \\ \text{C2} & \text{C3} & \text{C4} & \text{C4} \\ \text{C1} & \text{C2} & \text{C4} & \text{C4} \\ \text{C2} & \text{C3} & \text{C4} & \text{C4} \\ \text{C3} & \text{C4} & \text{C4} & \text{C4} \\ \text{C4} & \text{C5} & \text{C4} & \text{C5} \\ \text{C5} & \text{R}_1 = \text{R}_2 = \text{C0} \\ \text{C6} & \text{R}_1 = \text{R}_2 = \text{C0} \\ \text{C1} & \text{C1} & \text{C1} & \text{C1} \\ \text{C1} & \text{C1} & \text{C1} \\ \text{C1} & \text{C1} & \text{C1} \\ \text{C1} & \text$$

nucleus exhibits dienophilic character which is rather unusual since furan usually acts as a diene² in Diels - Alder reactions. The dienophilic character of furan has been mainly exhibited in connection with fused ring systems such as benzofuran and xanthotoxin^{3,4}. The reactivity of (I) toward addition to the furyl residue is supported by the ready formation of the adducts (IV, R=CH₃, C₂H₅, <u>iso</u>-C₄H₉, or C₆H₅) when it is allowed to react with the corresponding 2-dicyanoalkylfurans. The latter compounds are

readily obtained by the action of Grignard reagents on (IIa).

In contrast to the above mentioned 2-vinylfurans, it is found that the quinone adds both to the furyl residue and the vinyl side chain of the unsaturated cyanoester (IIf) to give the diadduct (V). This is rather unusual since it is known⁵ that o-quinones do not usually add to α,β -unsaturated ketones, nitriles and similar compounds.

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